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## Liquid Crystals

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# Pyrazolo[1, 2-a]pyrazoliumolates as novel structure elements in liquid crystals [1]

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### Pyrazolo[1, 2-a]pyrazoliumolates as novel structure elements in liquid crystals [1]

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A series of pyrazolo[1, 2-a]pyrazoliumolates of type 3, 4, and 5 was prepared. Compound  $4(R^3 = CH_3)$  shows liquid crystalline properties (C 158.5 N 162 I).

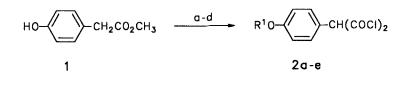
The overwhelming majority of compounds which exhibit thermotropic liquid crystal (LC) properties contain aryl and/or cycloalkyl groups [2, 3]. Heterocyclic compounds seemingly have been used to a less extent [4, 5]. In order to gain insight into the behaviour of LCs with highly dipolar structure elements [6], we tried to use cross-conjugated mesomeric betaines [7] as novel building blocks. In this paper, the first example of a pyrazolo[1,2-a]pyrazoliumolate [8, 9, 10] with LC properties is described.

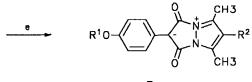
At the outset of our work it was still unknown whether *two* core elements would be sufficient for LC behaviour. Therefore 2-aryl substituted betaines of type **3** were prepared as shown in the scheme. Methyl 4-hydroxyphenylacetate was O-alkylated, transformed to malonic acid derivatives of type **2** and finally condensed with 3-alkyl-2,4-dimethylpyrazoles [11] to give **3a-e** [12]. Optical investigations showed that these betaines do not exhibit any LC behaviour. The elongation of **3f** [13], which could be prepared from **3b** by treatment with TiCl<sub>4</sub>/CH<sub>2</sub>Cl<sub>2</sub>, with an aromatic ester moiety proved to be successful. The esterification of **3f** with 4-*n*-butoxybenzoic acid yields **4**  $(R^3 = CH_3)$  [14] which shows LC behaviour albeit in a narrow region: on melting at 158.5°C a nematic phase is observed [15]; at 162°C the isotropic phase appears. On cooling down the range of the nematic phase could be extended to 142°C.

2, 3	$R^1$	<i>R</i> <sup>2</sup>	mp( <b>3</b> )/°C
a	C <sub>6</sub> H <sub>13</sub>	C <sub>6</sub> H <sub>13</sub>	111
b	$(CH_3)_2CH$	$C_{8}H_{17}$	126
с	$C_6 H_{13}$	$C_{8}H_{17}$	115
d	$C_{8}H_{17}^{13}$	$C_{8}H_{17}$	110
е	$C_{12}H_{25}$	$C_{8}H_{17}^{1}$	102
f	Ĥ	$C_8H_{17}$	178
g	$C_6H_{11}CO$	$C_{8}^{8-17}$	102

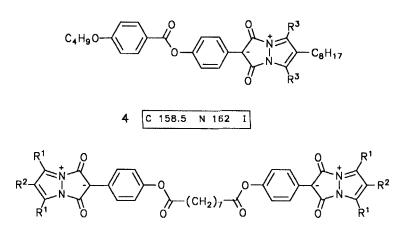
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3a-e



5a,b

 $a:R^1=CH_3, R^2=C_8H_{17}; b: R^1=C_5H_{11}, R^2=C_4H_9$ 

a: alkylbromide/K<sub>2</sub>CO<sub>3</sub>/acetone; b: OC(OEt)<sub>2</sub>/Na; c: KOH/ CH<sub>3</sub>OH; d: SOCl<sub>2</sub>; e: 3-alkyl-2,4-dimethylpyrazole/Et<sub>3</sub>N (alkyl: C<sub>6</sub>H<sub>13</sub>, C<sub>8</sub>H<sub>17</sub>)

The extension of these investigations to other compounds of type  $3 (C_5 H_{11} \text{ versus } CH_3)$  and twins of type 5 (prepared from bismalonic acid derivatives and pyrazoles) showed that these compounds did not exhibit LC behaviour. Whether cross-conjugated mesomeric betaines and other dipolar heterocycles (as pyrimidiniumolates [16, 17]) are advantageous or detrimental for LCs remains to be explored.

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- [12] All new compounds described in this communication have been fully characterized by IR, UV, <sup>1</sup>HNMR, <sup>13</sup>CNMR and high resolution mass spectrometry. For details see Supplementary Material which has been deposited in the British Library. (Copies of this material which comprises 9 pages may be obtained from the British Library, Lending Division, by quoting the number SUP 16524 according to the procedure described at the end of this issue.)
- [13] 3f: red crystals; mp 178°C (ethyl acetate-petrol ether); UV(CH<sub>3</sub>CN) λ (lg ε) 208 (4·58), 227 (4·51), 267 (4·75), 440 (3·07); <sup>1</sup>H NMR(CDCl<sub>3</sub>): δ 0·9 (t, 3 H, CH<sub>3</sub>), 1·25 (m, 12 H, CH<sub>2</sub>), 2·3 (t, 2 H, CH<sub>2</sub>), 2·50 (s, 6 H, CH<sub>3</sub>), 5·15 (s, 1 H, OH), 6·8 (d, 2 H), 7·9 (d, 2 H).
- [14] 4: orange red needles; mp 158·5°C (ethyl acetate); UV(CH<sub>3</sub>CN) λ (lg ε) 209 (4·53), 271 (4·64), 415 (2.78); <sup>1</sup>H NMR(CDCl<sub>3</sub>): δ 0·87 (6 H, CH<sub>3</sub>), 1·2–1·8 (m, 16 H, CH<sub>2</sub>), 2·33 (t, 2 H, CH<sub>2</sub>), 2·53 (s, 6 H, CH<sub>3</sub>), 4·03 (t, 2 H, OCH<sub>2</sub>), 6·95 (d, 2 H), 7·15 (d, 2 H), 8·13 (2 d, 4 H).
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